

Orbital Angular Momentum

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$$[\hat{L}_i, f(\hat{r})] = [\hat{L}_i, \hat{r}] f'(\hat{r}) = 0$$

for any function f of r .

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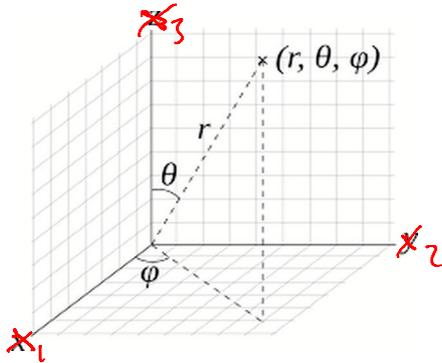
In other words, \hat{H} , \hat{L}_i and \hat{L}^2 all commute with one another.

This is an important and powerful result. Given any 3D quantum system, we can find a basis of simultaneous eigenfunctions of \hat{H} , \hat{L}^2 and \hat{L}_3 .

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We can translate the definitions of \hat{L}_i to spherical polars. We have

$$x_1 = r \sin \theta \cos \phi, \quad x_2 = r \sin \theta \sin \phi, \quad x_3 = r \cos \theta. \quad (7.51)$$



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We thus obtain

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We can also obtain

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We can thus seek simultaneous eigenfunctions of the form $Y(\theta) \exp(im\phi)$, since $\hat{L}_3 \exp(im\phi) = \hbar m \exp(im\phi)$. As ϕ is defined modulo 2π , we need $e^{im(\phi+2\pi)} = e^{im\phi}$, so $e^{i2m\pi} = 1$ and m is an integer.

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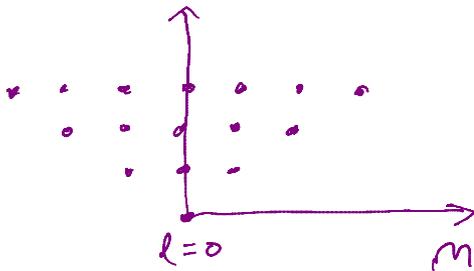
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$$-\hbar^2 \frac{d}{dw} \left((1 - w^2) \frac{dY}{dw} \right) - \left(\lambda - \frac{m^2}{1 - w^2} \right) Y = 0. \quad (7.60)$$

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$$P_{l,m}(\theta) = (\sin \theta)^{|m|} \frac{d^{|m|}}{d(\cos \theta)^{|m|}} P_l(\cos \theta). \quad (7.61)$$

(up to normalisation. Note that the solutions for $-m$ are proportional to those for m , for given l .)

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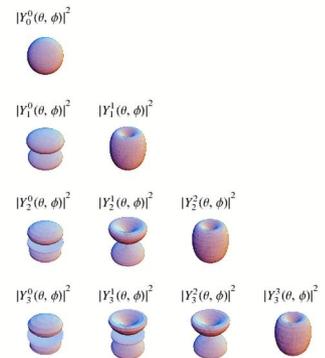
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For plots of some spherical harmonics see e.g. mathworld.wolfram.com/SphericalHarmonic.html.



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Solving the 3D Schrödinger equation for a spherically symmetric potential

The time-independent SE is

$$-\frac{\hbar^2}{2M}\nabla^2\psi + V(r)\psi = E\psi. \quad (7.62)$$

Recall that in spherical polar coordinates

$$\begin{aligned}\nabla^2 &= \frac{1}{r^2}\frac{\partial}{\partial r}\left(r^2\frac{\partial}{\partial r}\right) + \frac{1}{r^2\sin\theta}\frac{\partial}{\partial\theta}\left(\sin\theta\frac{\partial}{\partial\theta}\right) + \frac{1}{r^2\sin^2\theta}\frac{\partial^2}{\partial\phi^2} \\ \hat{L}^2 &= -\hbar^2\left(\frac{1}{\sin\theta}\frac{\partial}{\partial\theta}\left(\sin\theta\frac{\partial}{\partial\theta}\right) + \frac{1}{\sin^2\theta}\frac{\partial^2}{\partial\phi^2}\right).\end{aligned} \quad (7.63)$$

So we have

$$-\hbar^2\nabla^2 = -\hbar^2\left(\frac{\partial^2}{\partial r^2} + \frac{2}{r}\frac{\partial}{\partial r}\right) + \frac{1}{r^2}\hat{L}^2. \quad (7.64)$$

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We can thus rewrite the SE as

$$-\frac{\hbar^2}{2M}\left(\frac{\partial^2}{\partial r^2} + \frac{2}{r}\frac{\partial}{\partial r}\right) + \frac{1}{2Mr^2}\hat{L}^2)\psi(r, \theta, \phi) + V(r)\psi(r, \theta, \phi) = E\psi(r, \theta, \phi). \quad (7.65)$$

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So, we have a standard 1D radial Schrödinger equation for $\psi(r)$, with the modified potential $V(r) + \frac{\hbar^2 l(l+1)}{2Mr^2}$.

Solving the 3D Schrödinger equation for a spherically symmetric potential

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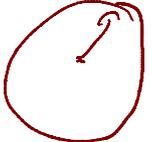
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This makes sense physically, since a state ψ with $\langle \underline{L} \rangle_{\psi} \neq \underline{0}$ by definition has a nonzero vector associated with it, which breaks spherical symmetry.

$$\begin{aligned} \psi(r) &= C \\ \psi(\theta) &= C \end{aligned}$$


$$\langle \underline{L} \rangle_{\psi}$$


Degeneracies

$$-\frac{\hbar^2}{2M}\left(\frac{d^2}{dr^2} + \frac{2}{r}\frac{d}{dr}\right)\psi(r) + \left(\frac{\hbar^2}{2Mr^2}l(l+1)\right) + V(r)\psi(r) = E\psi(r). \quad (7.66)$$

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The values of E for which this equation is solvable clearly may depend on l but not on m . As there are $(2l + 1)$ possible values of m , each energy level would have degeneracy $(2l + 1)$, assuming there are no further degeneracies.

Solving the 3D Schrödinger equation for a spherically symmetric potential

(f. 1D case: ground state
lowest energy state has even parity.  reflection symmetric pot^d.

Theorem

The ground state (i.e. lowest energy bound state) solution of the 3D Schrödinger equation for a spherically symmetric potential must have $l = m = 0$ and is thus spherically symmetric.

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I.e., the eigenstates ψ_i are all spherically symmetric solutions. We can thus write $\psi(r) = \sum_i c_i \psi_i(r)$ for some constants c_i such that $\sum_i |c_i|^2 = 1$.

Solving the 3D Schrödinger equation for a spherically symmetric potential

$$E = \int_{r=0}^{\infty} \psi^*(r) \left(-\frac{\hbar^2}{2M} \left(\frac{d^2}{dr^2} + \frac{2}{r} \frac{d}{dr} \right) \right) \psi(r)$$

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As we saw in discussing Eqn. (7.26), we see that the ansatz $\psi(r) \approx \exp(-br)$ means that the two asymptotically largest terms cancel. This again suggests trying an ansatz of the form $\psi(r) = f(r) \exp(-br)$, for a power series $f(r)$.

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a divergent and unnormalisable wavefunction, which is physically unacceptable.

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This is what we found previously for $l=0$

$$b = \frac{a}{2N} \quad n = 1, 2, 3, \dots$$

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$$E = -\frac{Me^4}{32\pi^2\epsilon_0^2\hbar^2} \frac{1}{N^2}.$$

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In fact, the true degeneracy of the N th energy level of the hydrogen atom in a full non-relativistic quantum mechanical treatment is $2N^2$: the extra factor of 2 arises from an intrinsically quantum mechanical degree of freedom, the electron *spin*, which has no direct classical analogue.

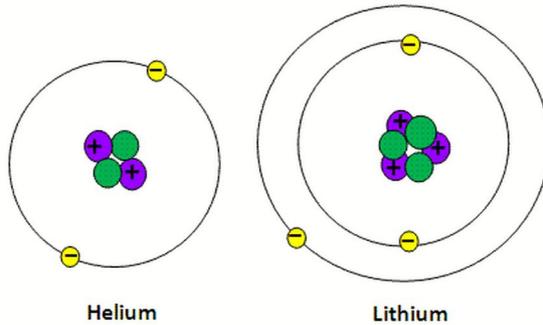
Towards the periodic table

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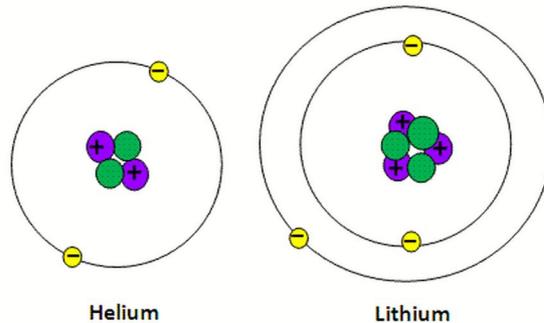


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If we ignore this temporarily, we can obtain solutions of the form

$$\psi(\mathbf{x}_1, \dots, \mathbf{x}_Z) = \psi_1(\mathbf{x}_1) \dots \psi_Z(\mathbf{x}_Z), \quad (7.78)$$

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where the ψ_j are rescaled solutions for the hydrogen atom (the nucleus has charge $+Ze$ instead of $+e$). The energy is just the sum

$$E = \sum_{i=1}^Z E_i. \quad (7.79)$$

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However, we also need to allow for the *Pauli exclusion principle*, which implies that no two electrons in the same atom can be in the same state.

So the lowest overall energy state is given by filling up the energy levels in order of increasing energy, starting with the lowest.

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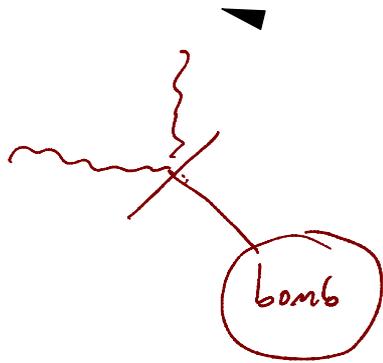
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Atomic #	Symbol	Name	Weight
1	H	Hydrogen	1.008
2	He	Helium	4.002602
3	Li	Lithium	6.94
4	Be	Beryllium	9.0121...
5	B	Boron	10.81
6	C	Carbon	12.011
7	N	Nitrogen	14.007
8	O	Oxygen	15.999
9	F	Fluorine	18.998...
10	Ne	Neon	20.1797
11	Na	Sodium	22.989...
12	Mg	Magnesium	24.305
13	Al	Aluminium	26.981...
14	Si	Silicon	28.085
15	P	Phosphorus	30.973...
16	S	Sulfur	32.06
17	Cl	Chlorine	35.45
18	Ar	Argon	39.948

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"good" bomb will explode if photon hits trigger

"dud" bomb will not explode - the photon just reflects.

Your problem: you have a stock of good + dud bombs,
and you don't know which is which.
Want to identify some good bombs.

Towards the periodic table

Allowing for the twofold degeneracy arising from spin, as above, we have $2N^2$ states in the N th energy level. This gives us an atom with a full energy level with $Z = 2, 10 = 8 + 2, \dots$ for $N = 1, 2, \dots$; these are the elements helium, neon, \dots . The elements with outer electrons in the 1st and 2nd energy levels fill out the corresponding first two rows of the periodic table. The analysis gets more complicated as atoms get larger, because electron-electron interactions become more important, and this qualitative picture is not adequate for the third and higher rows of the periodic table.

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